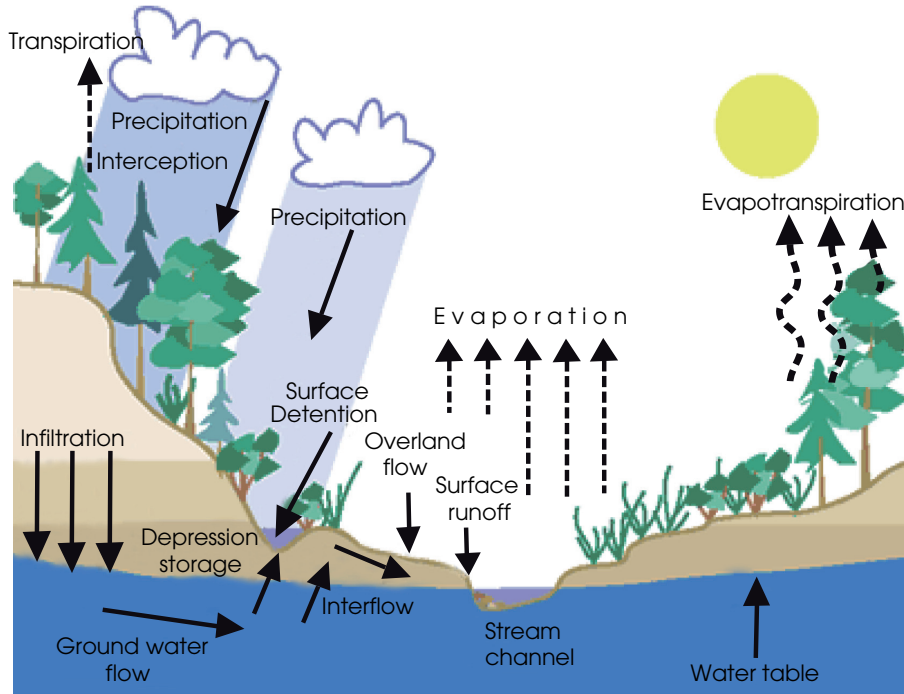
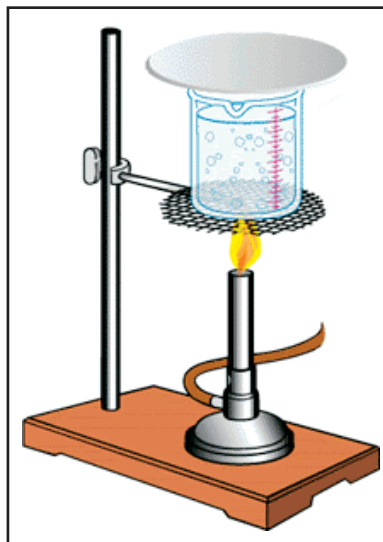


# Phase Changes



**Evaporation as water leaves ocean, Condensation as water forms clouds, Freezing as water goes from clouds to snow, and Melting as water goes from snow-capped mountains to run off.**



**Boiling**

# Mass and Weight On the Earth and the Moon

Mass equals  
54 kilograms



Weight = 120 pounds

Mass equals  
54 kilograms



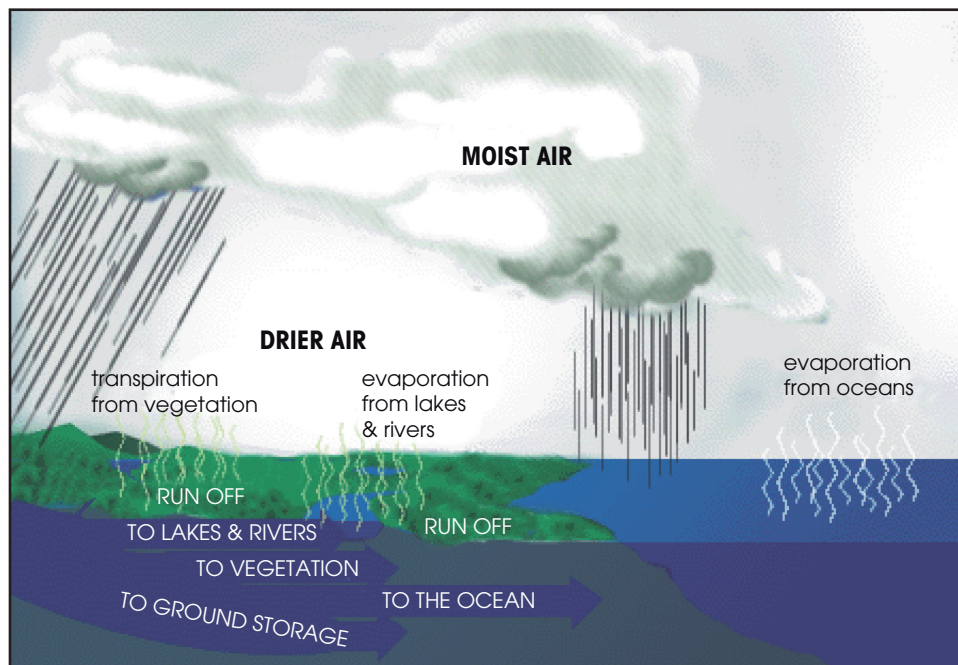
Weight = 20 pounds

# Chemical Changes and Physical Changes

**Chemical:** Forest fire burning



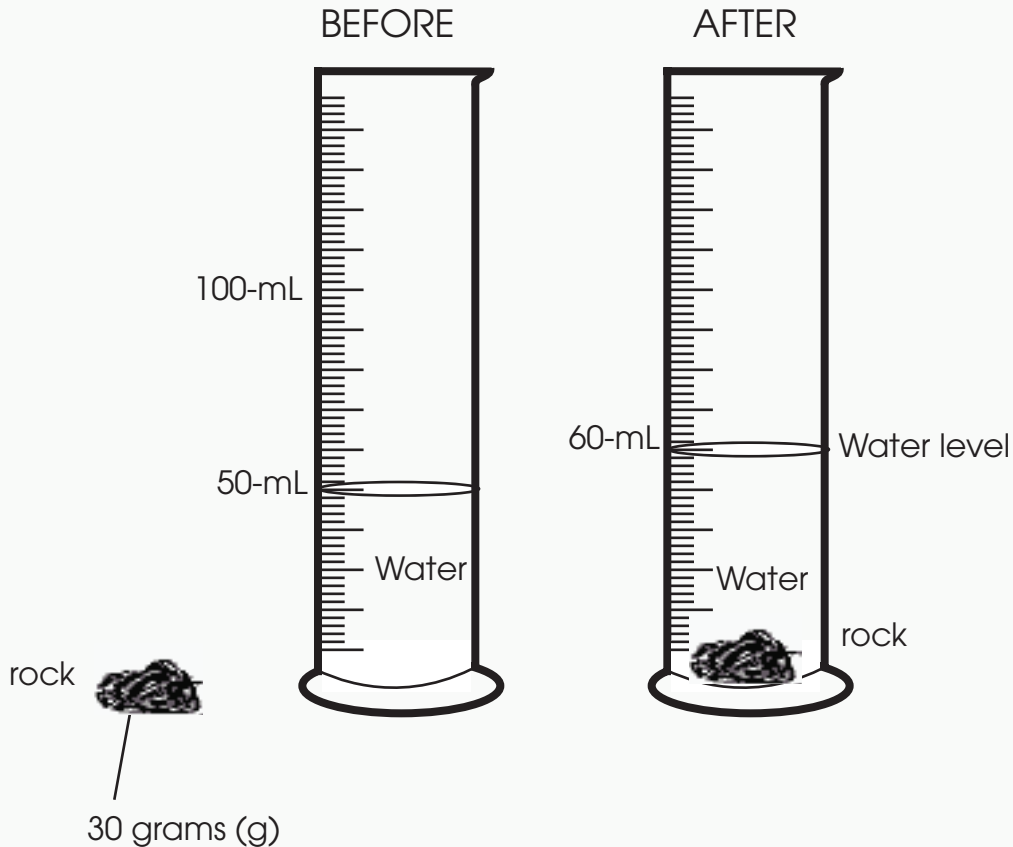
**Physical:** Water evaporating from oceans



# Finding Density



This is the **displaced liquid** method of finding density.



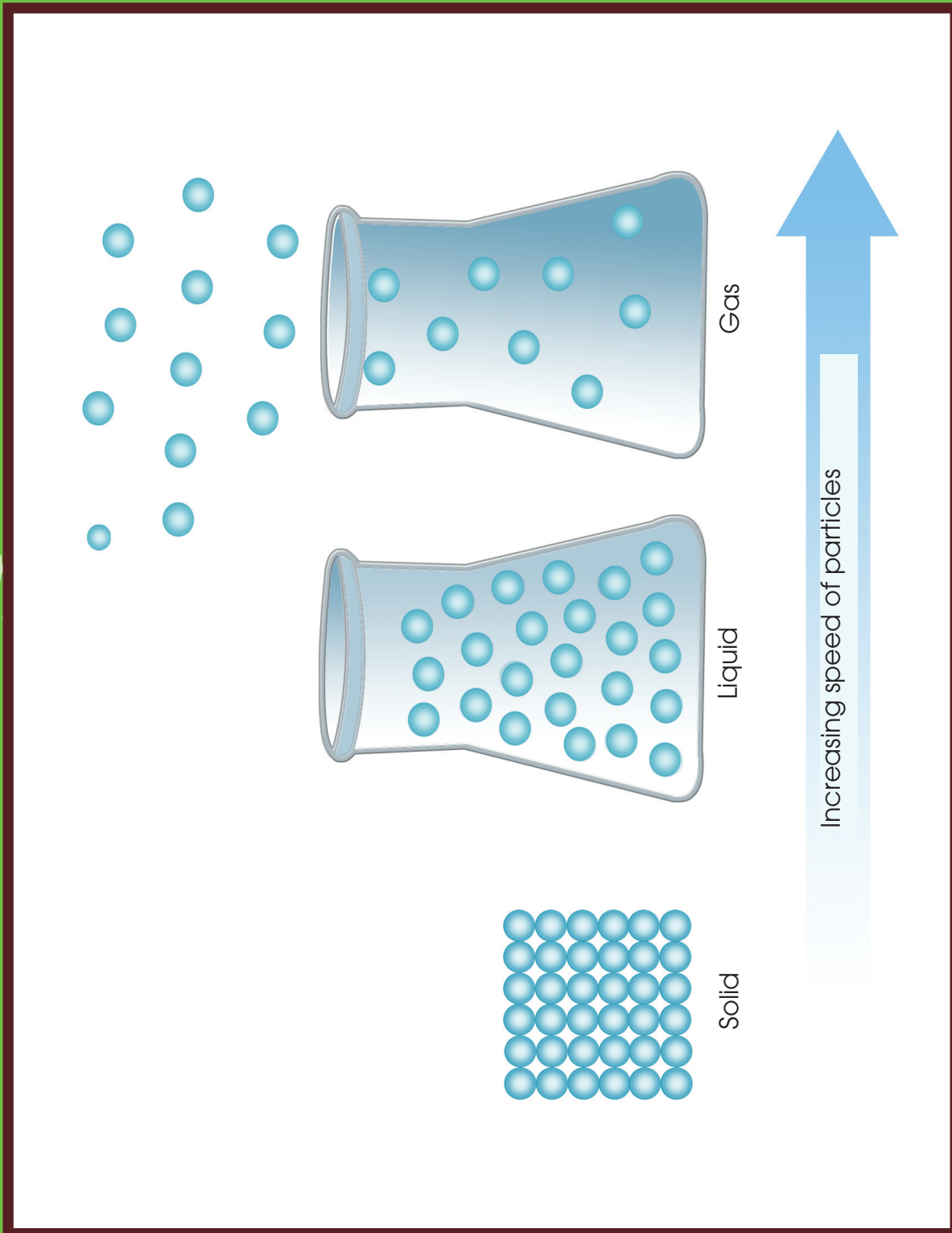
Mass of rock = 30 grams.

Volume of rock = 60 milliliter (mL) - 50 mL = 10 mL.

Density of rock =  $30 \text{ g} / 10 \text{ mL} = 3.0 \text{ g/mL}$

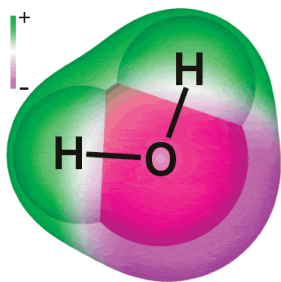


# Particles In Solids, Liquids and Gases

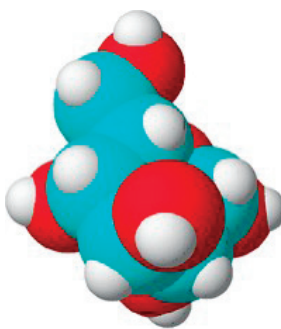


# Particles In Two Kinds of Mixtures

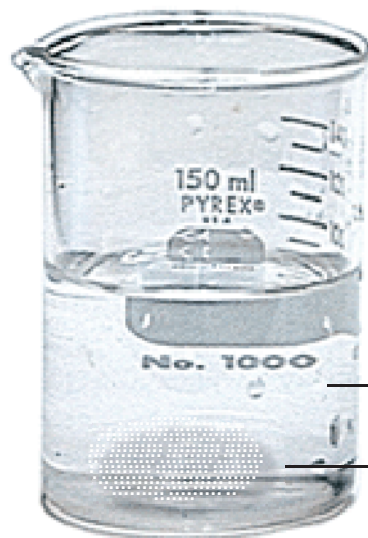
## Sugar particles in solution



Water molecule



Sugar molecule



— water particles

— sugar particles

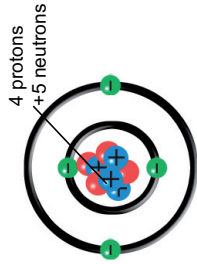


Sand grain

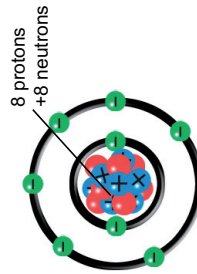


Sugar grain

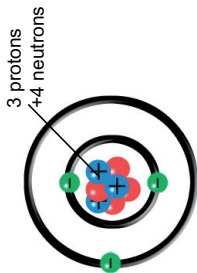
# Atomic Models



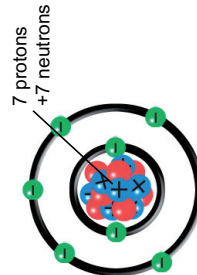
**Beryllium**



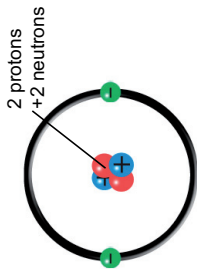
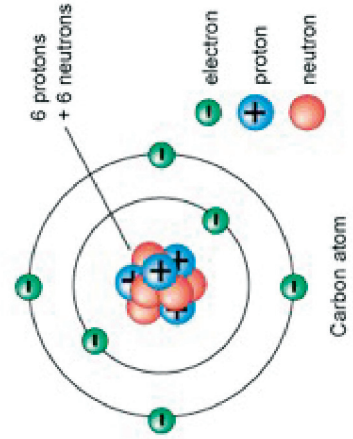
**Oxygen**



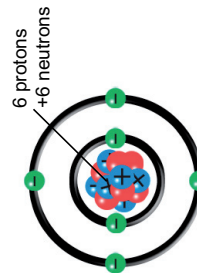
**Lithium**



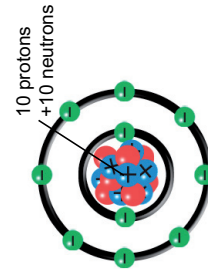
**Nitrogen**



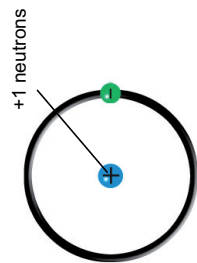
**Helium**



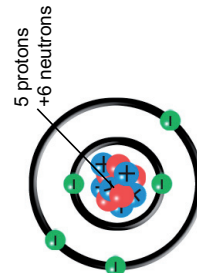
**Carbon**



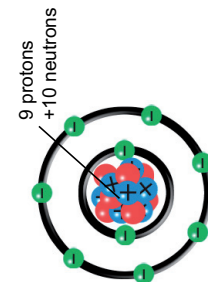
**Neon**



**Hydrogen**



**Boron**

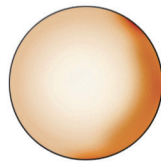


**Fluorine**

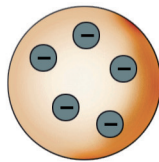
# History of the Atomic Model



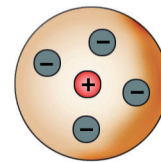
Dalton's atom



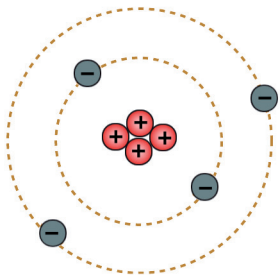
Thomson's plum-pudding atom



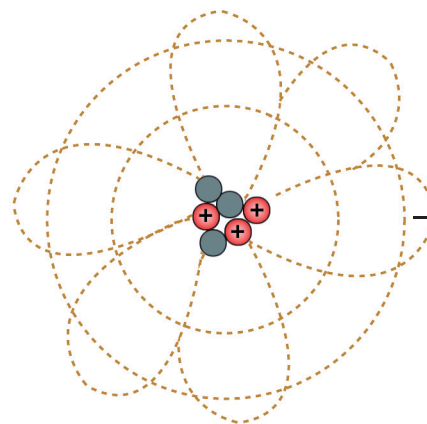
Rutherford's atom



Bohr's planetary atom



Current orbital atom



— Electron clouds

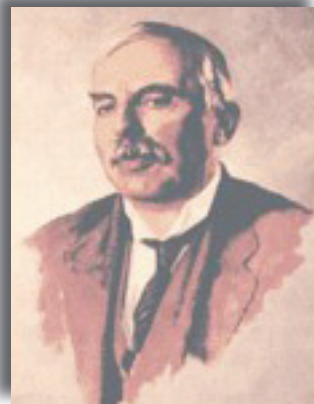
Current electron cloud model proposed in 1926.



Dalton  
1803



Thomson  
1897



Rutherford  
1909



Bohr  
1913



# Elements and Compounds



**OXYGEN**



**SALT**



**BAKING SODA**



**SUGAR**

# Mendeleev's Periodic Table



Group	I	II	III	IV	V	VI	VII	VIII
Period								
1	H=1							
2	Li=7	Be=9.4	B=11	C=12	N=14	O=16	F=19	
3	Na=23	Mg=24	Al=27.3	Si=28	P=31	S=32	Cl=35.5	
4	K=39	Ca=40	?=44	Ti=48	V=51	Cr=52	Mn=55	Fe=56, Co=59 Ni=59
5	Cu=63	Zn=65	?=68	?=72	As=75	Se=78	Br=80	
6	Rb=85	Sr=87	?Yt=88	Zr=90	Nb=94	Mo=96	?=100	Ru=104, Rh=104 Pd=106
7	Ag=108	Cd=112	In=113	Sn=118	Sb=122	Te=125	J=127	
8	Cs=133	Ba=137	?Di=138	?Ce=140				
9								
10			?Er=178	?La=180	Ta=182	W=184		O=195, Ir=197 Pt=198
11	Au=199	Hg=200	Tl=204	Pb=207	Bi=208			
12				Th=231		U=240		



Mendeleev

Mendeleev's Periodic Table



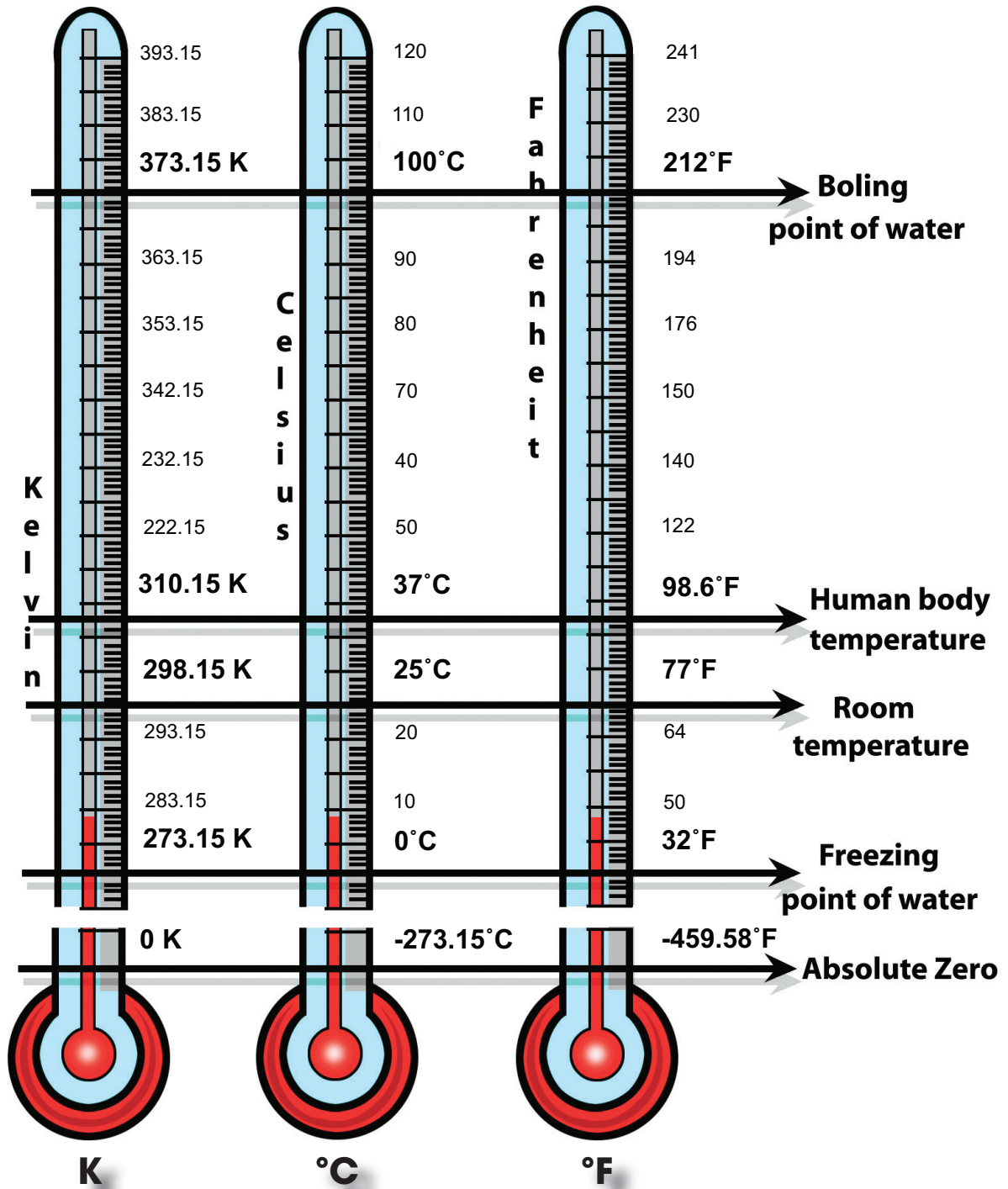




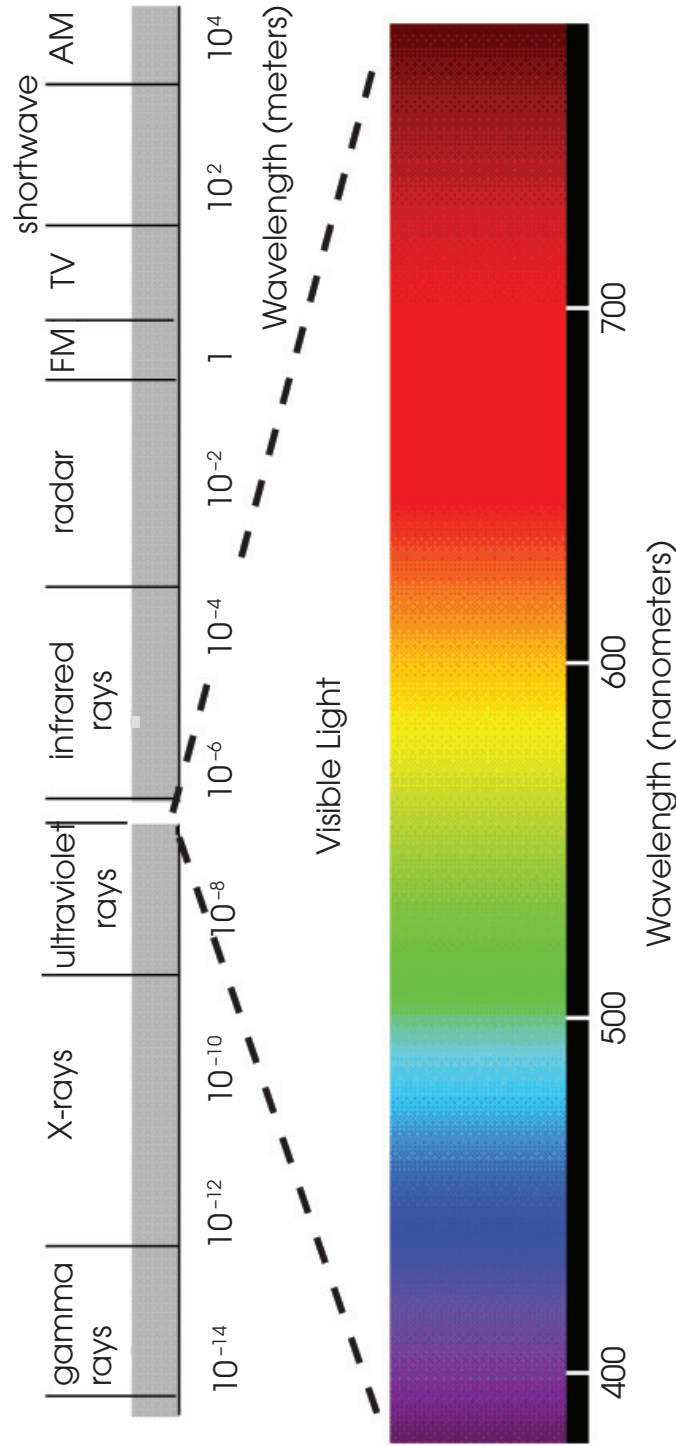
# Potential and Kinetic Energy on a Rollercoaster



# Temperature Scales



# Electromagnetic Spectrum

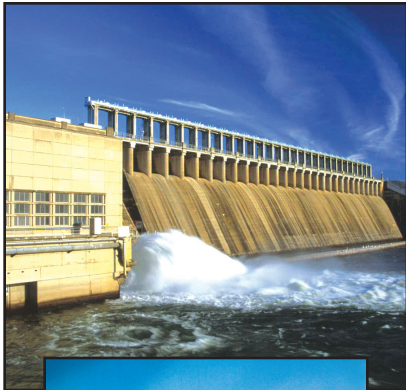


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[Visual Stimulus](#)



# Energy Sources



Hydroelectric Power

WOOD (BIOMASS)



WIND



WAVE



SOLAR

## RENEWABLE ENERGY SOURCES



COAL

GAS



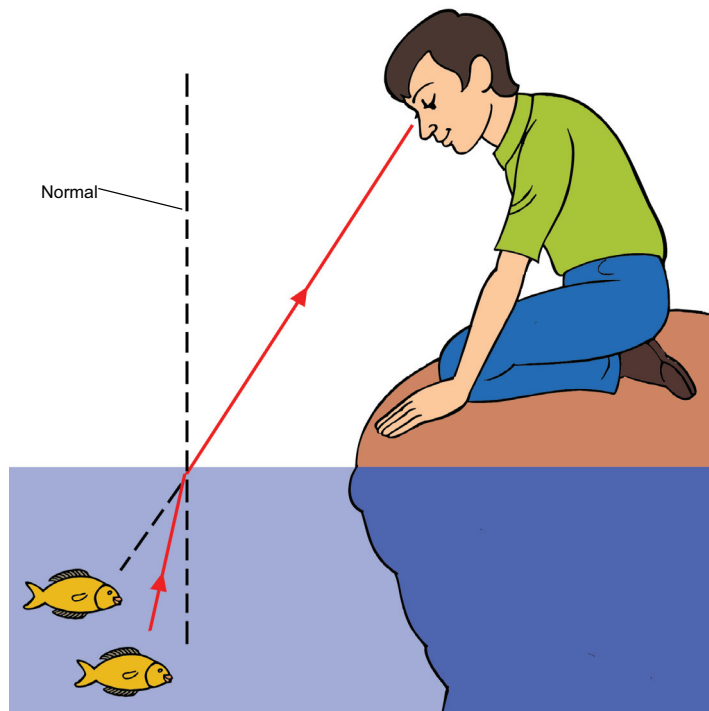
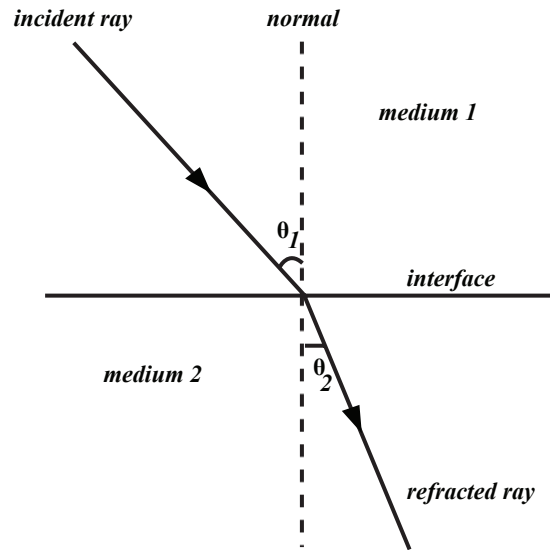
NUCLEAR



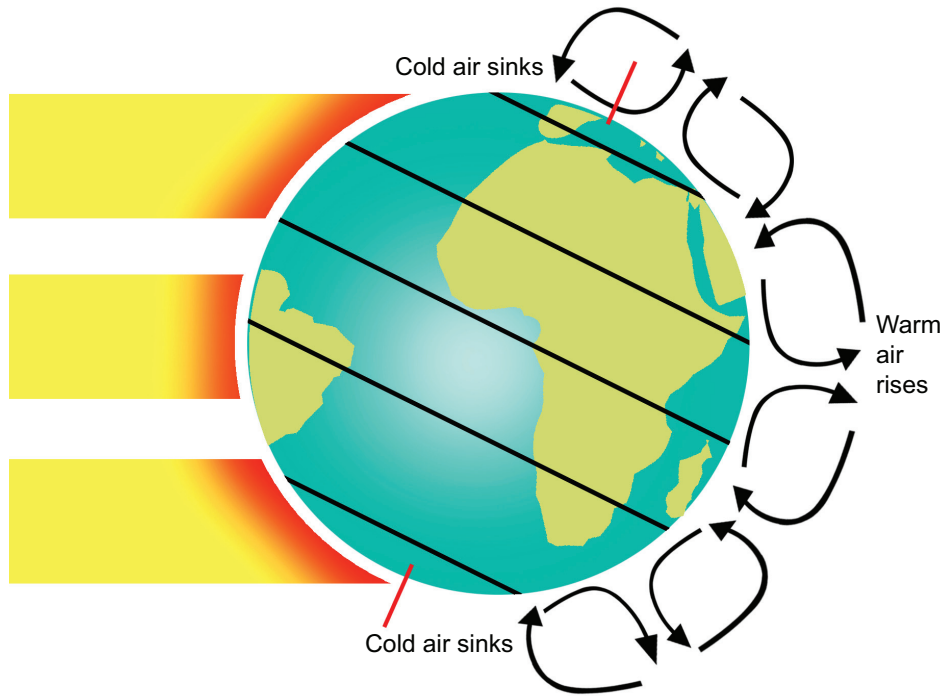
## NONRENEWABLE ENERGY SOURCES



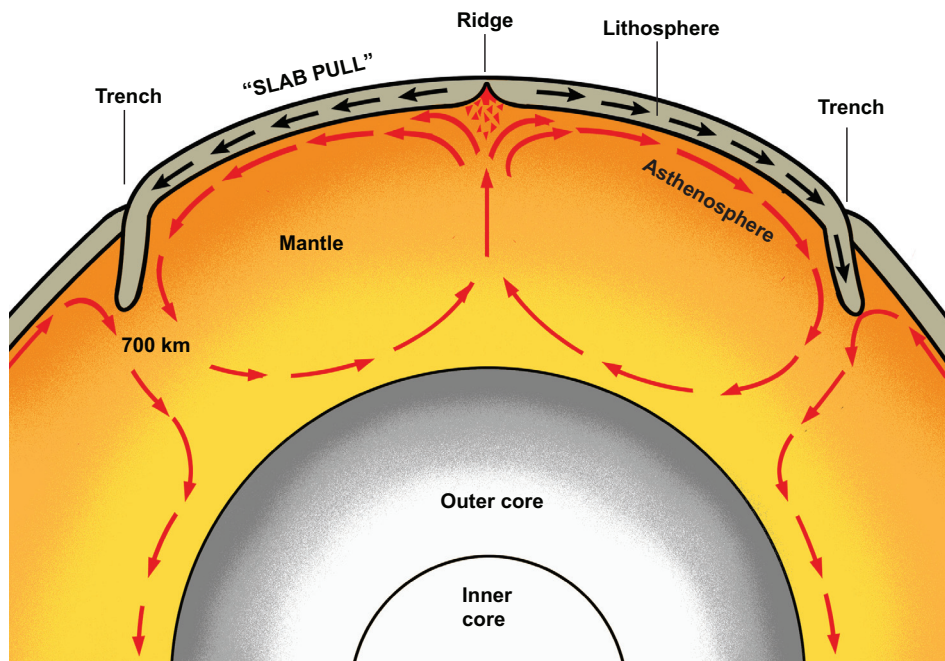
# Refraction



# Convection Currents in the Atmosphere and in the Mantle



**CONVECTION IN THE ATMOSPHERE**



**CONVECTION IN THE MANTLE**