MATTER AND ITS CHANGES

UNIT OVERVIEW

In this fast-paced unit students discover that "matter" matters. An engaging array of activities combined with interesting worksheets compliments the concepts brought forward in the overhead student notes. Relating the study of matter, atoms, and molecules to the "real world" is essential. Students delight as they learn about DNA fingerprinting and why a grade two class eating pop and chocolate bars is important to the study of chemistry. Optional activities add flexibility and an element of fun to the unit. Finally, a unit on atoms and molecules that will not give students that "glazed eye - dead fish" look.

STUDENT ASSIGNMENTS AND ACTIVITIES

- 1. Matter Matter Matters: Introductory Wordsearch
- 2. Matter and Its Three States Excited States (Worksheet)
- 3. Physical and Chemical Properties Get Physical (Classifying and Observing)
- 4. Amazing Atoms Amazing Atom Quizzz
- 5. Magnificent Molecules DNA: The Crimestopper Molecule (Worksheet)
- 6. Elements Elementary My Dear Watson (Crossword)
- 7. Metal/Non-metal Elements Poster Making Activity
- 8. Changes In Matter (Physical Changes) Which Freezes Faster Hot Water or Cold?
- 9. Changes In Matter (Chemical Changes) (Worksheet)

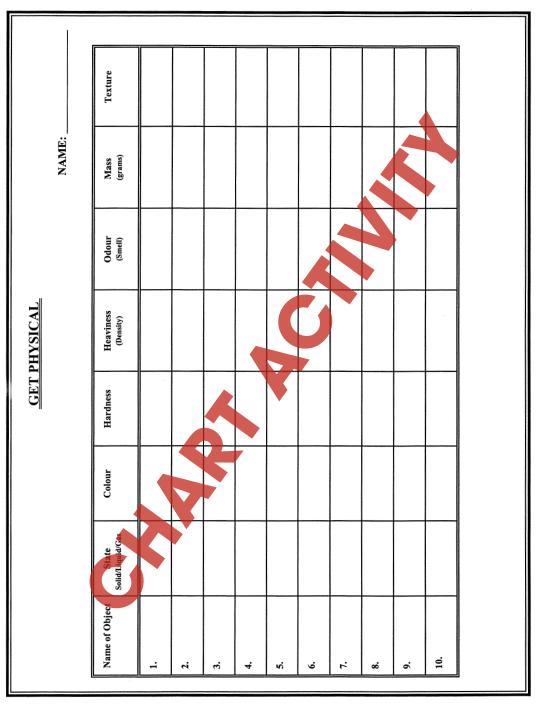
OPTIONAL ACTIVITIES

These optional activities lend a degree of freedom to the structure of the unit. Teachers can use them as they see fit as enrichment for individuals or fun activities for the entire class. Generally, the optional assignments do not fit as well into the specific concepts discussed in the unit but with a bit of imagination, they can usually be integrated without difficulty.

- 1. Giant Bubbles
- 2. Can Steel Float?
- 3. Needle Homemade Compass
- 4. Baking Soda And Vinegar
- 5. Crayon Melting Art Activity
- 6. Making Sandpaper
- 7. A Cool Evaporation Garden

OVERHEAD NOTES

The overhead notes provide a framework of knowledge and concepts upon which the activities in the unit are based. The student notes work best when photocopied onto overhead transparencies but can be written on the board, dictated or handed out as photocopies, depending on time considerations.



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DNA: THE CRIMESTOPPER MOLECULE

NAME:

The DNA Fingerprint

Molecule Of Life Inside every cell of every living thing is a molecule called deoxyribonucleic acid. (It is no wonder the name was shortened to DNA!) DNA molecules make up the chromosomes found inside the center of each cell and carry all the information or "genes" that the animal or plant will need to grow. DNA is like an instruction booklet or a "blueprint" telling how to build a living thing. Half of a person's DNA is inherited from their mother and half from their father.

Big! Big! Big!

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Like other molecules, DNA is a combination of two or more elements - except there is a difference. While salt (NaCl) is made by joining sodium atoms (Na) with chlorine atoms (Cl), and water molecules (H₂O) are made by joining two hydrogen atoms (H) to a single oxygen atom (O), DNA combines thousands or millions of atoms of carbon, oxygen, nitrogen phosphorus and hydrogen. DNA is a complicated molecule.

The Double Helix
In addition to being large, DNA
molecules also have a special shape. The atoms are joined together into two, long spiral chains called the "Double Helix".

Recently, forensic (crime) scientists have started to use DNA to help solve crimes. As scientists learned more about DNA they discovered that although most humans have very similar DNA, one person's DNA is a tiny bit different from anyone elses. (except for identical twins) They also learned that the DNA molecules were all the same throughout a person's body. For example the DNA found in a persons blood cells would be the same as the DNA in their bone cells hair cells. In this way DNA

could be used like a sort of

"fingerprint" to tell people apart.

DNA And The Fight On Crime Today, forensic scientists use special tests to help solve crimes. If a person is suspected of committing a crime, they can be make to give a sample of their DNA (blood or hair) to the police. If the police have found any evidence at the scene of a crime that has DNA in it - like hair, blood drops or skin cells - they can check it against the suspect's DNA. If the DNA is the same, then the police are more likely to get a conviction and put the person in jail.

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NAME: _ Label the parts of the atom List important points about each. Atom Nucleus **Electrons** When two things combine to form a brand new substance a chemical change has taken place. Explain why electrons are so important when chemical changes happens.

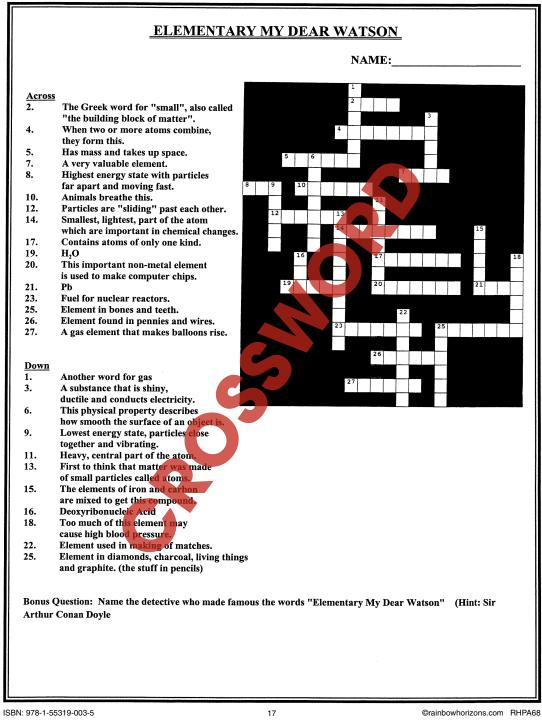
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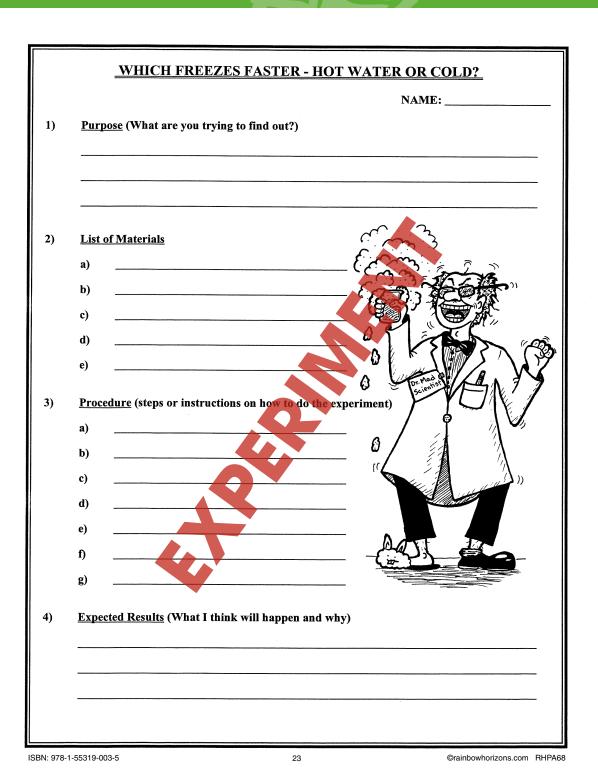
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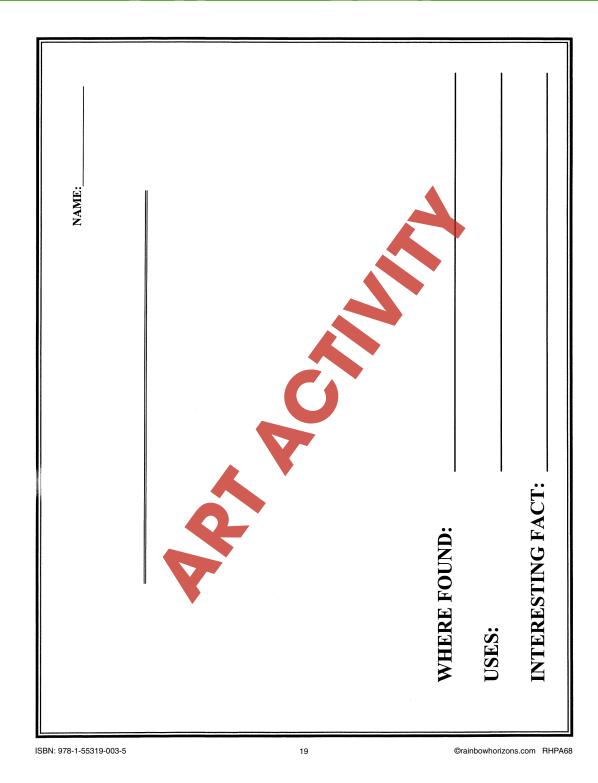
THE AMAZING ATOM QUIZZZ

What is the nu	n Full Sentences)
What is the pu	rpose of DNA in a living thing?
How is a mole	cule of DNA different from a molecule of salt
Describe the sl	hape of a DNA molecule.
Why is it impo	ortant to forensic scientists that everyone's DNA is a little bit different
~ "	
Describe how	DNA evidence might be used to help convict a criminal.
Describe how	DNA evidence collected at a crime scene could become contaminated.

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	CHEMICAL AND PHYSICAL CHANGES
	NAME:
1.	What is the difference between a chemical change and a physical change?
2.	What is a compound?
3.	List the things that tell a person if a chemical change has happened.
	1)
	3)
4.	Five raisins are placed in a jar
	of water containing vinegar and
	two tablespoons of baking soda. What do you think will happen?
	Wall do you tallak wat appear
_	Positive and the second
5.	Explain a possible reason for your answer in question 4.
I	

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MATTER MATTERS

NAME:

C	H	E	M	I	C	A	L	C	\mathbf{H}	A	N	\mathbf{G}	\mathbf{E}	R
Q	Y	R	\mathbf{C}	O	\mathbf{N}	D	\mathbf{U}	C	T	O	R	G	\mathbf{U}	\mathbf{F}
I	E	\mathbf{V}	A	P	O	R	A	T	I	O	\mathbf{N}	O	A	G
Q	M	E	T	A	L	O	I	T	P	A	P	В	\mathbf{H}	\mathbf{S}
\mathbf{E}	G	M	O	L	\mathbf{V}	\mathbf{U}	A	L	\mathbf{H}	A	D	E	\mathbf{U}	\mathbf{Y}
I	Y	E	M	C	\mathbf{U}	S	\mathbf{H}	C	\mathbf{V}	D	R	E	G	D
N	T	\mathbf{L}	L	H	N	\mathbf{Z}	${f L}$	D	\mathbf{E}	U	L	X	I	\mathbf{N}
M	A	T	T	E	R	A	I	T	T	C	P	\mathbf{U}	C	E
P	\mathbf{V}	I	D	M	C	L	A	X	\mathbf{U}	T	Q	\mathbf{J}	L	C
G	T	N	H	I	O	T	E	N	\mathbf{N}	I	\mathbf{U}	\mathbf{V}	E	S
C	0	\mathbf{G}	$\mathbf{S}_{\mathbf{r}}$	S	S	T	R	E	L	L	\mathbf{W}	O	W	T
C	L	Y	S	T	\mathbf{F}	\mathbf{A}	M	O	L	\mathbf{E}	C	\mathbf{U}	L	${f E}$
T	H	A	\mathbf{F}	R	E	\mathbf{E}	Z	I	N	G	D	\mathbf{C}	J	\mathbf{W}
P	M	J	\mathbf{S}	Y	\mathbf{L}	\mathbf{H}	G	P	W	S	W	0	R	F
F	X	В	K	E	K	O	A	X	H	S	Y	H	P	X

Find These Words:

ALCHEMIST ATOM CHEMICAL CHANGE CHEMISTRY CONDENSATION CONDUCTOR DUCTILE ELECTRONS ELEMENT
EVAPORATION
FREEZING
GAS
LIQUID
MASS
MATTER
MELTING

METAL
MOLECULE
NUCLEUS
PHYSICAL CHANGE
SOLID
STATE
TEXTURE
VAPOUR

V #1 - MATTER

Objectives and Activities

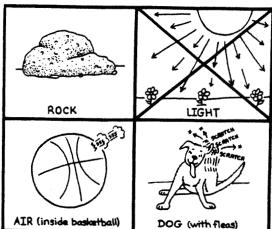
tudents learn and understand what matter is and are able to classify things as matter or on-matter by applying the definition.

tudents complete a wordsearch "Matter Matters" which helps to familiarize them with ocabulary introduced during the course of the unit.

d Teaching Strategies

egin by showing students the diagram on the first age of the overhead notes *without* giving them the efinition of matter or the topic of the unit.

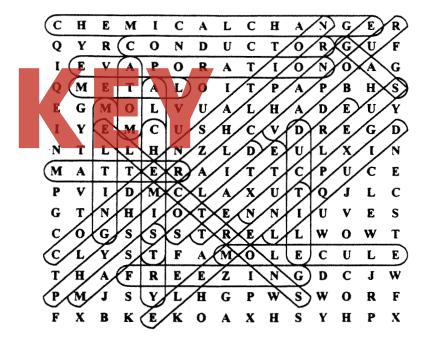
troduce the unit topic with that old familiar game f "One Of These Things Is Not Like The Others." Iopefully, some students will remember the words and tune to the song, adding a sense of realism to be activity)



lost students will likely guess correctly that light is "not like the others" but then ask the udents to explain why. This will lead into the topic and a discussion of what matter is. 'ut an "X" through light)

omrlete the notes with students answering a question about matter. Then, handout the ch. Clues are horizontal, vertical and diagonal but <u>not</u> inverted.

Matter Matters (Solution)



he publishers have declined to include the words to "One Of These Things Is Not Like he Others" as a result of threats posed by the notorious "Sesame Street Copyright iolation Police" and a wish not to confront Big Bird when his feathers are "ruffled."